## Y4 SCIENCE KNOWLEDGE - ORGANISER AUTUMN 1: WHAT ARE SOUNDS AND HOW DO WE HEAR THEM? THRESHOLD CONCEPT: PHYSICS – INVESTIGATE SOUND AND HEARING

Key Vocabulary				
vibration	A movement back and forth to create	vacuum	A space with nothing in it— not even air.	
	a sound.			
wave	A form that sound takes as it moves	sound proof	Something such a material that prevents	
	through air, water etc. Recorded on a		the passage of sound through it.	
	graph.			
volume	The loudness of a sound.	decibel	A unit measurement given to the loudness	
			or intensity of a sound.	
pitch	The quality related who whether	eardrum	The part of the ear that vibrates when	
	sounds are 'high' or 'low'.		receiving sounds.	
tone	A musical or vocal sound with	insulation	Material used to trap air and sound.	
	reference to its pitch, quality, and			
	strength.			
Sound is a type of <b>energy</b> . Sound is produced by		Super Scientis	t de la companya de la	

Sound is a type of **energy**. Sound is produced by vibrations, even when it is hard to see them. The vibrations travel through the air and are detected by our ears. Within the ear is an ear drum which vibrates and turns the vibrations into signals to the brain, which then 'hears' thecsounds.

The **loudness of a sound** depends on the size of the vibration: the bigger the vibration, the louder the sound. The greater the volume of air vibrating, the louder the sound will be. A large drum struck with the same force as a small drum will sound louder because the bigger drum can make more of the air move, simply by have a bigger 'skin' to vibrate. A vibrating tuning fork cannot be heard until the stem is placed on a table. This causes the table to vibrate very slightly, but there is a large volume of air in contact with it compared to the small volume of air in contact with the prongs of the tuning fork.

**Sound waves** spread out in all directions from the source of the sound. The front of the wave is in the shape of a sphere, which gets larger as the wave moves through the air. The energy from the source is spread out over the surface of the sphere.



As a scientist, da Vinci was interested in how sound moves through different materials. He is often credited with discovering that sound travels in waves, allowing Galileo to later on discover more properties of sound waves.



**Sound (or vibrations)** enters the ear through the ear canal. When sound waves reach our ear, it travels through the ear canal and hits the eardrum, causing vibrations. The eardrum sends these vibrations to three tiny bones in the middle of the ear (the hammer, the anvil and the stirrup) into the inner ear. The inner ear is known as the cochlea and is shaped a bit like a snail. There are thousands of tiny hair cells inside the cochlea. These hair cells change the vibrations into electrical signals that are sent to the brain through the hearing nerve. The brain tells you that you are hearing a sound and what that sound is.



#### Y4 SCIENCE KNOWLEDGE - ORGANISER AUTUMN 2: HOW DO WE CLASSIFY LIVING THINGS AND WHAT DANGERS FACE **THEM AND THEIR HABITATS?** THRESHOLD CONCEPT: BIOLOGY - INVESTIGATE LIVING THINGS

Key Vocabulary				
skeleton	The bones of the body form a framework called the skeleton. This framework supports and protects the softer tissues.	ecology	The study of organisms, and how living and non-living things interact with each other.	
environment	The environment refers to the surroundings in which life exists on earth.	characteristics	A distinguishing quality.	
classification	To arrange or group things into	cold-blooded	An animal whose body temperature	
key	categories depending on characteristics.		varies with the environment they are in, i.e. fish.	
habitat	A natural environment for any type of living organism.	exoskeleton	An external covering of the body found in some invertebrates such a as arthropods.	
species	A group of similar organisms that are able to reproduce.	warm blooded	Animals that maintain a regular body temperature	
vertebrate	An animal with a backbone.	invertebrates	An animal without a backbone	

Super Scientist

organisms.

botanist, physician, and

His work has influenced

many generations of prominent scientists, including Charles Darwin

and Gregor Mendel.

system for classifying

There are millions of different **plant** and **animal species** in the world. It's currently estimated that there are around 1.3 million animal species (of which 1 million are insects!) and 320,000 plant species. We use the word 'species' to describe a group of living things that are so similar.

Insects are the only invertebrates that are able to fly. This has played a major role in their success.

Living things are divided into groups, with members of each group having similar features. The obvious first grouping is whether something is an animal or a plant (or fungus or microbe).

# Why classify?

The process of classification makes it easier to identify a species when it is discovered, and to see whether it is an existing species or a new species. It can also help to see which species are closely related in evolutionary terms.

The animal kingdom can be divided into two broad groups based on whether they have a backbone (vertebrate) or not (invertebrate). The five groups of vertebrates are fish, amphibian, reptile, bird and mammal. The plant kingdom can also be divided into two groups: flowering and non-flowering plants.

Fish, reptiles and amphibians are cold-blooded. Amphibians have lungs, which allow them to spend a lot of time out of water, but they return to water to lay lots of jelly-like eggs. Reptiles have dry scaly skin and live on land, where they lay their eggs – these look a lot like bird eggs. Mammals and birds are warmblooded. Birds are covered in feathers and lav eaas with a hard shell. Mammals all have fur (or hair) - even if it is very fine, as in the case of whales and dolphins. They all give birth to live young and female mammals produce milk to feed them.

Carl Linnaeus was a Swedish zoologist who formalised the

Myriapods (meaning 'many legs') have long, thin bodies with many segments and a hard exoskeleton (exomeaning 'outside').

**Centipedes** have one pair of legs per body segment and can have between 20 and 300 leas.

Millipedes have two pairs of legs per segment and can have between 36 and 400 legs – not the million legs that you might think they have!

The **arachnids** (including spiders and scorpions) have four pairs of leas and a two-part body. They have a hard exoskeleton, which often forms lots of protective bristles.

# CLASSIFICATION OF ANIMALS



#### Y4 SCIENCE KNOWLEDGE - ORGANISER SPRING 1: WHAT ARE THE STATES OF MATTER AND CAN THEY CHANGE? THRESHOLD CONCEPT: CHEMISTRY – INVESTIGATE MATERIALS

Key Vocabulary			
solid	An object with a definite shape. The particles in a solid are tightly packed together, and cannot move about	temperature	A degree of hotness or coldness the can be measured using a thermometer.
liquid	A substance that flows freely but is of constant volume.	precipitation	Liquid or solid particles that fall from a cloud as rain, sleet, hail or snow.
gas	A substance which will expand freely to fill a whole container and has no fixed shape or volume.	heating	Heat is the transfer of energy from one object to another.
evaporation	The process of turning a liquid into a gas.	reversible	Materials can be changed back to how they were before the reaction took place.
condensation	The process of turning a gas into a liquid.	irreversible	A change is called irreversible if it cannot be changed back again.
particle	An extremely tiny piece of matter	freezing	A change of state from liquid to solid.
states of	Materials can be one of three states:	melting point	The temperature at which a given solid
matter	solids, liquids or gases. Some		will melt
	materials can change from one state to another and back again.		
A material may exist in <b>three states: solid, liquid, and</b> <b>gas.</b> The particles of these states all behave differently.		Super Scientist John Dalton	1 Prog

The state that a material is in depends on the temperature. Water, for example, is in its solid state (ice) at 0°C or below, liquid state (water) between 0 and 100°C and, at temperatures of 100°C and above, water exists in the gas state, as steam.

Super Scientist	
John Dalton	
1766 – 1844	
In 1803 he proposed matter is	
made up of atoms that are	
indivisible and indestructible.	



SOLID











RIGID FIXED SHAPE FIXED VOLUME



NOT RIGID NO FIXED SHAPE NO FIXED VOLUME

When a sample of a material is in the **solid state**, you can hold it in your hands. You can form it into a pile. It is not easy to change the shape of a material in the solid state. You may question this as a sponge is a solid. You can squash a sponge, but it is the air you are 'squeezing', not the sponge itself.

When a material is in the **liquid state**, you cannot hold it in your hands. It forms a pool, not a pile. Liquids take the shape of the bottom of the container they are in. Another misconception would be about sand being a solid but it runs through your fingers. You need to consider each grain as a tiny solid.

In the **gas state**, a material escapes from an unsealed container. It spreads out to fill all the space available, and takes the shape of the entire container. When a sample of a material melts, it turns into a liquid, because heat has been applied. **Evaporation** can happen at any temperature. The higher the temperature, the faster a material evaporates. Evaporation is speeded up if moving air carries the particles away from the surface of the liquid. It is evaporation that dries wet clothes.

**Boiling** occurs throughout a material in the liquid state. Bubbles rise to the surface, where they escape to the air. A material can change from the gas state to the liquid state by **condensing**. Condensing happens at any temperature below the boiling point, but happens most readily at cold temperatures. These processes are linked in the water cycle.

Freezing, Melting and Boiling points of water



There are bonds between particles in a solid; as temperature increases, these bonds are partially overcome as the particles absorb energy and solids can change into liquids; with an increase in temperature the particles become even more energetic and the bonds are overcome entirely so the liquid changes into a gas.

## Y4 SCIENCE KNOWLEDGE - ORGANISER SPRING 2: WHY ARE OUR TEETH AND DIGESTIVE SYSTEM IMPORTANT? THRESHOLD CONCEPT: BIOLOGY – UNDERSTAND ANIMALS AND HUMANS

Key Vocabulary					
mouth	The place where food is chewed and	nutrients	A substance that offers nourishment to the		
	mixed with a clear liquid called saliva.		body. We need these to be healthy.		
tongue	An organ, or body part, in the mouth.	stomach	An organ which helps to break down food.		
	It is made up of a group of muscles.				
teeth,	Teeth are the hardest substances in	canine	The sharp pointy teeth in mammals'		
	the human body. Besides being		mouths.		
	essential for chewing, the teeth play				
	an important role in speech.	• •			
oesophagus	A muscular tube connecting the	Incisor	leeth useful for biting because they cut		
	throat and the stomach.		snarpiy.		
absorb	To take comothing in	molar	Tooth used for grinding food		
small intesting	The small intestine breaks down food	producer	The name given to a living thing that		
SITION INCOMINE	from the stomach and absorbs much	producer	produces its own food, rather than		
	of the nutrients from the food.		consuming another living thing.		
larae	Food material is turned into faeces, or	salivary	A aland which produces saliva (spit) to		
intestine	solid waste.	gland	help you digest your food.		
Humans are <b>or</b>	nnivores, meaning we eat both plants	Super Scientis	st Contraction of the second se		
and animals, a	nd our teeth have evolved to suit our	William Beaur	mont		
diet.		1785 – 1853			
		Beaumont wo	as a U.S. army		
Our digestive s	ystem is made up of organs that take in	surgeon, the	first person to observe		
food, including	our mouth and teeth, which start off	and study hu	man digestion as it		
digestion as a i	mechanical process, to then digest it	occurs in the	stomach.		
chemically to e	extract energy and nutrients, and expel	Most of the e	xperiments were		
the remaining v	waste.	conducted by tying a piece of			
Food contains	large complex chemicals such as	through the hole in a patient's			
carbohydrates	proteins and fats. To be of use to the	stomach.			
body, they mu	st be broken down into smaller	Every few hou	urs. Begumont would		
chemicals:		remove the fo	remove the food and observe how		
		well it had be	een digested.		
Digestion starts	in the mouth. Teeth provide	All living thing	as need energy to survive. Plants are able to		
mechanical breakdown of the food, then saliva		use the energy from the Sun to produce their own food.			
moistens food	so that it slides down the <b>oesophagus</b>	Animals are unable to make their own food so have to			
into the stoma	ch. The stomach is a bag of muscle that	eat other living things to get their energy.			
breaks up tooc	by churning it around. It also contains				
nyarochioric a	cia, which kills off bacteria in the food,	Other animals have adapted to eating only other			
and enzymes, which further break down		animals and get their energy from meat. These animals			
breakdown of	the food After a few hours in the	are known as predators, with the animals that they eat			
stomach food travels down the duodenum (small		known as prev, whether they are herbivores or carnivores.			
intestine), when	re it is broken down further and sends	and the proy, whether may dre herbivores of currivores.			
the nutrients around the body in the blood. The solid		Esophagus			
waste such as	fibre that can't be digested continues				
along into the large intestine, where water is removed.		Liver —			
			Stomach		
100 m		Gall			
		Bladder -	Pancreas		
Mouth Salivary					
	giands				
Oesophagus		Intestine			
			Small Intesting		
Intestine			Intestine		
		Appendix -			
			19-013-		
Gall bladder	Stomach				
Pancreas			Rectum		

#### Y4 SCIENCE KNOWLEDGE - ORGANISER SUMMER 1: WHAT ARE ELECTRICAL CIRCUITS AND HOW DO THEY WORK? THRESHOLD CONCEPT: PHYSICS – UNDERSTAND ELECTRICAL CIRCUITS

Key Vocabulary			
appliance	A machine in your home that you use	series	A looped circuit where the electricity flows
	to do a job such as cleaning or		from the positive to negative terminal of
	cooking.		the battery.
battery	The main source of energy that	motor	An electrical machine that converts
power	provides a voltage, which allows the		electrical energy into mechanical energy.
	current to flow through.		
main power	The term used to refer to the electricity	switch	A component within an electrical circuit,
	supply from power stations to		which enables the flow of electricity to be
	households.		turned on and off.
circuit	A complete path around which	break in	A break along the circuit, which stops the
	electricity can flow.	circuit	current travelling.
insulator	Materials that do not allow electricity	conductor	Materials which allow electricity to flow
	to pass through them with ease		through them with ease.
cell	Part of a battery.	wire	Wire used to carry electricity
battery	A device that can make electricity	bulb	A device used to create light (illumination)

Electricity is the most useful form of energy. It can be transformed into other forms of energy relatively easily. It makes things turn using motors, heats and lights up places like our homes, and produces sound in loudspeakers. Most mains electricity is produced in power stations and carried to users by overhead power lines.

Batteries contain chemicals, which react in a special way to produce an electric current. Voltage indicates the amount of energy delivered by a source of electricity. The voltage of the most common household batteries varies from around 1.5V to 12V, Mains electricity in this country is 230V. Overhead power cables carrying sufficient supply for thousands of users can carry voltages as high as 400,000V.

There are two types of circuit. A series circuit has all its components wired into one simple circuit: all the components are one after another, as in a series on television. A parallel circuit is one with different branches which behave like mini-circuits and can work independently of each other.



**Super Scientist Maria Telkes** 1900 - 1995Maria is a famous scientist

who made lots of discoveries around solar power. She is best known for creating the first house built with a heating system that ran completely on solar energy.



Metals are good conductors of electricity. Most nonmetals do not conduct electricity. They are **insulators**.

Conventional bulbs contain a filament made from wire. As electrons flow through the wire, they encounter resistance. When a wire is very thin, it has a large resistance. The electrons get hot as they try to move through it (just as people do going through a small doorway!) and we can feel this. If it gets very hot, it glows - as in the filament of a light bulb.

When investigating the changing of components in a circuit, the brightness of a bulb depends on the current or number of electrons passing through it. The more bulbs you have, the slower the electricity flows because the battery 'runs out of push', so the electrons flow more slowly, due to the resistance through the wires, and the bulbs, so the dimmer the light.

# Electrical Circuit Symbols





buzzer







lamp







closed switch

### Y4 SCIENCE KNOWLEDGE - ORGANISER SUMMER 2: WHAT IS THE SCIENCE BEHIND BUBBLES? THRESHOLD CONCEPT: CHEMISTRY – INVESTIGATE MATERIALS

Key Vocabulary				
diluted	A solution with a lot of water	concentration	The ratio of water (solvent) to substance (solute).	
solute	A substance that has been dissolved in another substance	bubble	A thin sphere of liquid enclosing air or another gas.	
solution	A mixture of one or more solutes dissolved in a solvent.	mixture	A substance made by mixing other substances together.	
solvent	The substance that does the dissolving.	carbon dioxide	Gas found in the air and in fizz	
concentrated	A solution with a small amount of water	sphere	A shape that's round like a ball.	

A **bubble** is a thin film of soapy water filled with gas. Most bubbles are filled with air, but you can use other gases, such as carbon dioxide.

The film that makes the bubble has three layers. A thin layer of water is sandwiched between two layers of soap molecules.



When you blow a bubble, the film expands outwards. The forces acting between the molecules of the bubble cause it to form the shape that encloses the most volume with the least surface area – in other words, a sphere. No matter what shape a bubble has initially, it will try to become a sphere. This is the shape that requires the least energy to achieve.

Bubbles pop all of a sudden because the water in the bubble mix simply evaporates. Glycerine/ glycerol gives the mixture extra strength, by making the soap layer thicker, which slows down the evaporation process.

Bubbles don't stay as spheres. When two bubbles meet, they merge their walls to minimise their surface area. If bubbles that are the same size meet, then the wall that separates them will be flat.

#### Super Scientist Robert Hooke

In 1672, the English scientist Robert Hooke made a mass of bubbles, which was created in a soap solution by blowing into it through a glass tube.



Yeast is a single celled fungus that is 50% protein and is a rich source of vitamin B. It is always floating around in the air but is often too small for the human eye to see. Yeast grows best on foods with a lot of sugar or starch, such as fruit. Like every living thing, yeast produces carbon dioxide as it grows and requires oxygen to survive. It also respires, and is the ingredient, which makes champagne fizzy and makes bread rise.



Aero bars are made from chocolate in a liquid state that's just on the verge of solidifying. The exact recipe for making the bubbles so evenly spaced is a closely guarded commercial secret. Maltesers are made in two stages, which again are a closely guarded secret. The inner crunchy part is made from small pellets of a doughlike mixture. This is placed in a low-pressure container, which forces the bubbles in the mixture to expand, making the pellets larger. The pellets are then coated with chocolate and polished.